

5'-NOR ANHYDROVINBLASTINE

PROTOTYPE OF A NEW CLASS OF VINBLASTINE DERIVATIVES

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(Received in UK 19 April 1979)

Abstract—The reaction medium obtained after applying the modified Polonovski reaction to anhydrovinblastine *N*₆-oxide 4 led, after hydrolysis, to 5'-nor anhydrovinblastine 10a.

The most fruitful approach for the preparation by partial synthesis¹ of the complex antitumour alkaloids of the vinblastine 1 group found in *Catharanthus spp.*, took advantage of several stereoelectronic factors for directing functionalization of carbons C₂₀ and/or C₁₅ of anhydrovinblastine ($\Delta^{15(20)}$ 20'-deoxyvinblastine) 2 or other intermediates.²⁻⁴

The modified Polonovski reaction⁵ applied to anhydrovinblastine *N*₆-oxide 4 appeared to be well suited to prepare intermediates as the conjugated immonium salt 5.

In contrast with the regioselectivity generally observed in the case of tetrahydropyridines,⁵ the modified Polonovski reaction applied to the *N*₆-oxide 4 did not lead only to the conjugated immonium 5 (path a, Scheme 1) but also to a fragmentation product of the tryptamine chain (path b, Scheme 1) giving rise to new seco C₃-C₆ derivatives 7 depending on the nucleophile being used.⁶ We wish to disclose the preparation of this new class of vinblastine derivatives, the first example of which being the 5'-nor anhydrovinblastine 10a.

Thus, the product of the reaction of anhydrovinblastine *N*₆-oxide 4 with trifluoroacetic anhydride was treated, after evaporation of the excess of reagent, with a mixture of water and tetrahydrofuran. In addition to a mixture of polar products, probably constituted by immonium salts and/or corresponding carbinol amines, the only product isolated was 5'-nor anhydrovinblastine 10a (yield: 27%) (Scheme 2). Treatment of this polar mixture with sodium cyanoborohydride in acidic tetrahydrofuran led to the recovery of anhydrovinblastine 2 (35%) and the formation of 20'-deoxy leurosidine 3 (18%), these two products being formed by reduction of the conjugated immonium salt 5 (Scheme 1).

The formation of 5'-nor anhydrovinblastine 10a can be easily rationalised: addition of water on the bisimmonium salt 6 and subsequent loss of formaldehyde followed by a nucleophilic displacement (with or without intervention of a retro-Mannich process) of the substituent on carbon C₆ (Scheme 2).

The UV spectrum of 5'-nor anhydrovinblastine 10a shows an indole-dihydroindole chromophore while the CD curve is analogous to those of vinblastine derivatives, indicating that there is probably no change in chirality on C atoms C₁₄ and C₁₆.^{1,7}

In the PMR spectrum (250 MHz) of compound 10a, the signals corresponding to the functional groups of vin-

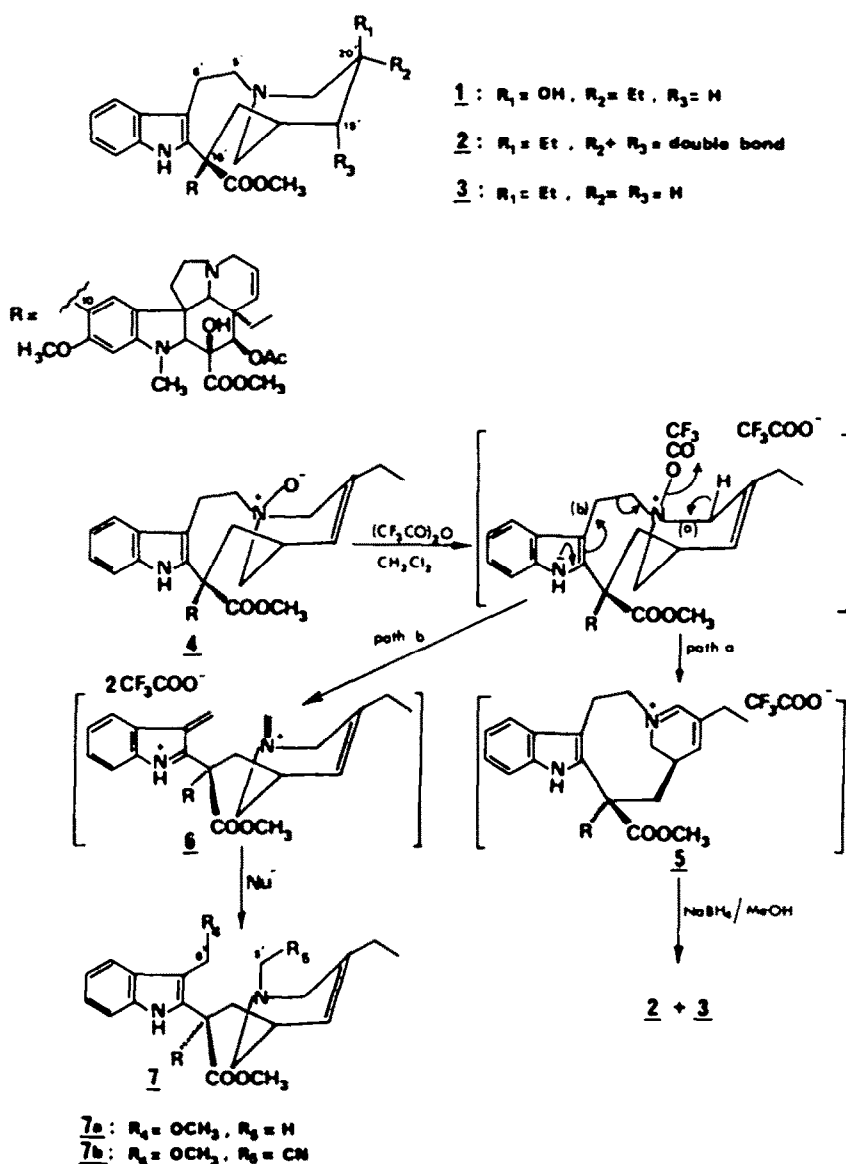
doline are all present indicating that the vindoline part of the molecule has been preserved. The signal at 5.76 ppm of the ethylenic proton at C₁₅ can be compared to the corresponding signals in 7 (seco C₃-C₆) which lies at ≈ 5.0 ppm; this striking difference disfavours structure 10b for this compound (Scheme 2). Two doublets of one proton each appear at 4.58 and 4.41 ppm respectively; decoupling experiments allow to assign them to an A-B system (*J* = 14 Hz) attributed to C₆-H₂. The chemical shifts for these protons are in agreement with a gramine-type methylene⁸ which are modified depending on whether the molecule is carbonated or not.

In the CMR spectrum, signals corresponding to the vindoline part are assigned in comparison with spectra of other vinblastine type derivatives.⁹ These signals are generally well resolved in contrast with those corresponding to the "nor ibogane" part of the molecule. This fact could be due to either the conformational flexibility of the large membered ring⁹ or to protonation of *N*₆ during registration of the spectrum (CO₂). For this nor ibogane part of the molecule, in addition to signals attributed at high field to the Et chain and OMe, five CH₂ (53.6, 47.0, 44.1, 35.2 and 27.7) and one CH (28.2) were observed; one CH₂ is missing in comparison with anhydrovinblastine 2. In addition, a signal at 53.6 ppm, superimposed to the signal corresponding to C₇, can be attributed to a gramine-type CH₂.¹⁰

The mass spectrum of 5'-nor anhydrovinblastine 10a exhibits a very small molecular ion (*M*⁺ *m/e*: 778) and other fragments at *m/e*: 808, 794, 792, 748, 734 and 598 (100%). The presence of ions characteristic of the vindoline part at *m/e* 282, 135, 122 and 121 indicates again that this part of the molecule is indeed intact.¹¹ Intermolecular thermal transmethylation followed by Hofmann elimination are well known to occur during mass spectroscopy of compounds related to vinblastine 1.¹² High resolution measurements for ions at *m/e* 734 and 748 (778-CO₂ and 792-CO₂) indicate that these ions could be formed through an intra or intermolecular transmethylation process followed by the loss of a methoxy-carbonyl group.

These ions are shifted by three mass units in the corresponding 10d, derivative (C₁₆-CO₂CD₃) indicating that intermolecular transmethylation followed by loss of methoxy carbonyl on C₁₆ is not involved. On the other hand, using a chemical ionization technique the main fragment at high molecular weights appears at *m/e* 735 (734 + 1); therefore an intramolecular transmethylation followed by a decarboxylation seems to be the most likely hypothesis.

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Scheme 1.

However, lack of obvious molecular ion in the mass spectrum of compound **10a**, prompt us to undertake further experiments concerning this type of compound.

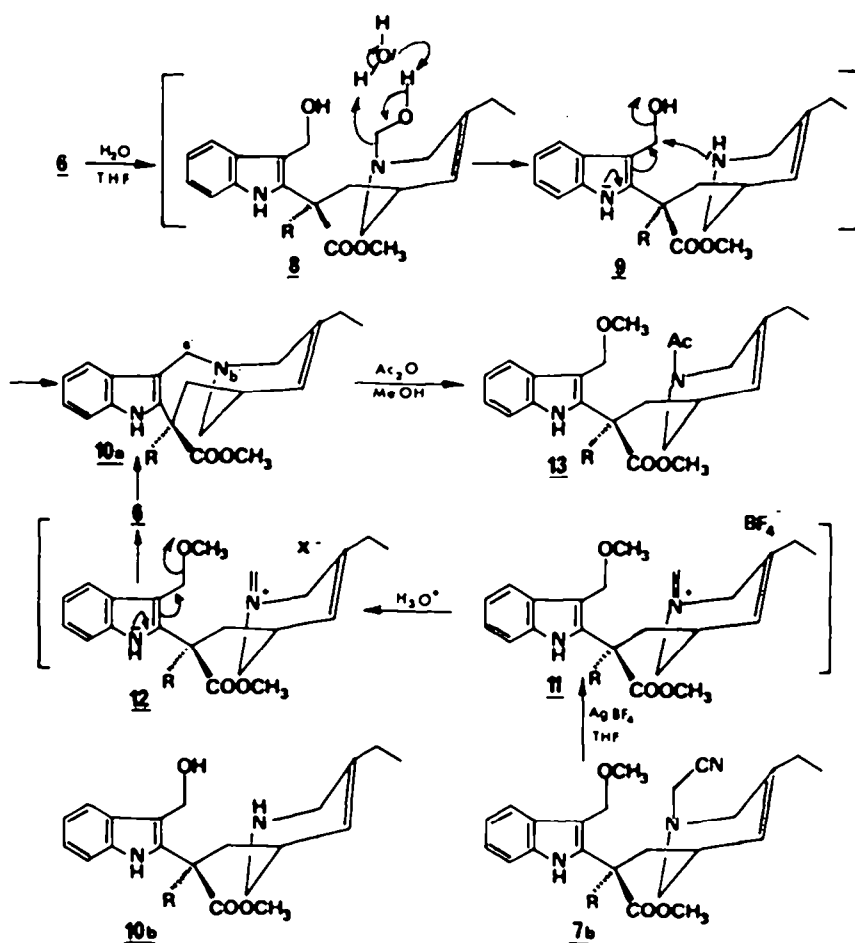
Thus, compound **7b**⁴ when treated with silver tetrafluoroborate in tetrahydrofuran and then with acidic water gave rise to 5'-nor anhydrovinblastine **10a**: silver salt regenerates the corresponding immonium salt **11** by decyanation and the acidic medium causes the elimination of the OMe group leading to the conjugated immonium salt intermediate **6** (or equivalent compound) (Scheme 2).

Conversely, 5'-nor anhydrovinblastine **10a** was treated by acetic anhydride in methanol⁸ giving rise to the corresponding N-acetylated compound **13** by fragmentation of the $\text{N}_6\text{-C}_8$ bond (Scheme 2).

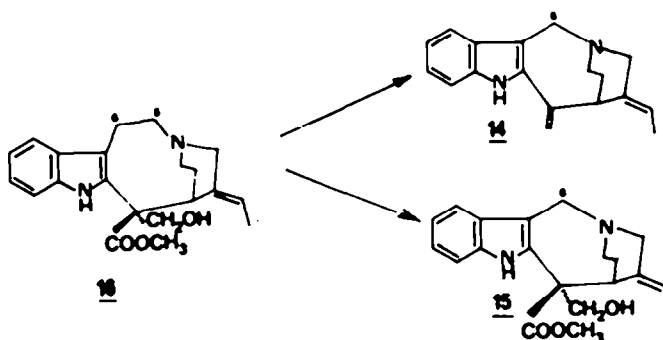
Structure of compound **10a** was definitely established by its clean N-oxidation with *m*-chloroperbenzoic acid a reaction which allows to exclude structure **10b** for this

compound. The PMR spectrum of the N_6 -oxide of 5'-nor-anhydrovinblastine **10a** was recorded at 400 MHz¹⁸ (Fig. 1). The AB system corresponding to $\text{C}_4\text{-H}_2$ is shifted at 4.91 and 4.68 ppm ($J = 13$ Hz) and signals corresponding to other protons on C_{21} and/or C_7 α to N_6 also appear at lower field.

Biogenetic implications of the fragmentation of the tryptamine chain induced by the modified Polonovski reaction,¹³ leading to C_5 -nor-alkaloids, postulated by Potier and Janot,¹⁴ were fully confirmed *in vivo* for apparicine **14**¹⁵ and *in vitro* for vallesamine **15**¹⁶ obtained from their common biogenetic precursor, i.e. stemmadenine **16** (Scheme 3). It is noteworthy that 5'-nor-anhydrovinblastine **10a** is the first example of a new class of antitumour compounds and could well be a natural product like anhydrovinblastine itself which was prepared *in vitro* in our laboratory before being isolated from *Catharanthus roseus*.¹⁷



Scheme 2.



Scheme 3.

EXPERIMENTAL

M.p.s were taken on a Kofler apparatus, optical rotations measured (CHCl_3 soln, g/100 ml) on a Perkin-Elmer 141 MC, IR spectra ($\nu \text{ cm}^{-1}$, CHCl_3) on a Perkin-Elmer 257, UV spectra [EtOH , λ_{max} , nm (ϵ)] on a Bausch and Lomb Spectronic 505, CD curves [EtOH , λ_{max} , nm ($\Delta\epsilon$)] on a Roussel-Jouan Dichrograph II. ^1H NMR spectra were obtained (CDCl_3 , Me_4Si , $\delta = 0$ ppm) from Varian T 60 or IEF 240 and 400 MHz¹⁸ and Camca 250 MHz spectrometers (coupling constants, J , are given in hertz; s, d, t, dd and m indicate singlet, doublet, triplet, doublet of doublet, and multiplet, respectively). Mass spectra were measured on an AEI MS 9 and MS 50. Preparative layer

chromatography (preparative tic) is performed with Kieselgel HF 254 Merck.

Anhydrovinblastine N_1 -oxide 4. *m*-Chloroperbenzoic acid (86 mg, 0.56 mmol) in CH_2Cl_2 (5 ml), was added at 0° to a stirred soln of 2 (340 mg, 0.43 mmol) in CH_2Cl_2 (10 ml) under argon. After 10 min, the mixture was poured into a sat of Na_2CO_3 aq (3 ml) extracted by CHCl_3 (50 ml), washed three times with brine (5 ml). After drying with Na_2SO_4 and filtration, the CHCl_3 extract was evaporated under reduced pressure. Pure 4 was obtained (346 mg, 100%). UV: 268, 289, 299, 310. MS m/e : 822, 820, 808 (M^{+}), 806, 792, 733, 669, 631, 610, 282, 136, 135, 122, 121 (100%). NMR (240 MHz): 9.80 (s, 1H, $\text{C}_1\text{-OH}$), 8.19 (s, 1H, $\text{N}_1\text{-H}$), 7.70

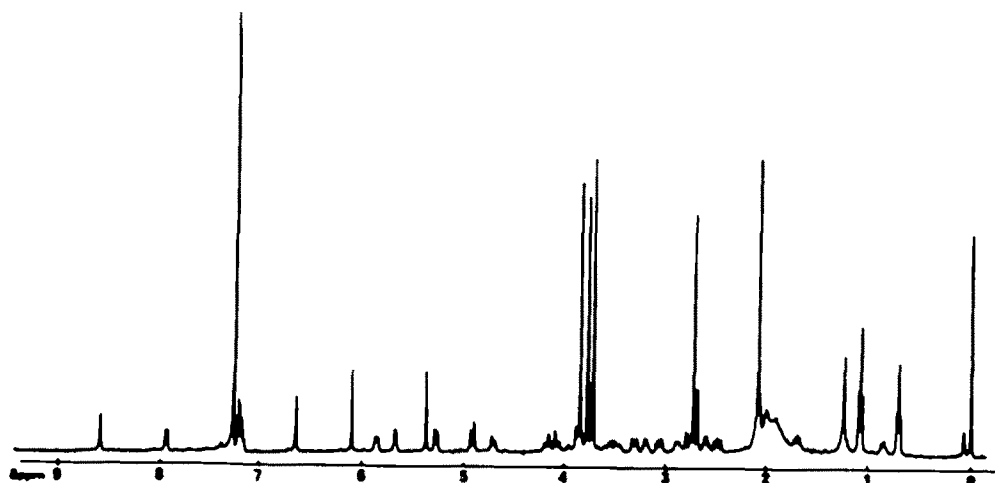


Fig. 1.

(d, 1H, $J = 7.5$, H aromatic indolic), 7.14 (m, 2H, H aromatic), 6.43 (s, 1H, C₉-H), 6.11 (s, 1H, C₁₇-H), 5.84 (dd, $J_{14,15} = 10$ and $J_{3,11} = 4.5$, C₁₅-H), 5.44 (1H, C₁₇-H), 5.41 (s, 1H, C₁₇-H), 5.27 (d, 1H, $J_{14,15} = 10$, C₁₅-H), 4.51 (m, 2H), 4.32 (d, 1H) and 3.98 (d, 1H) N₆-CH, 3.84, 3.78 and 3.69 (3s, 9H, C₁₁-OCH₃, C₁₈CO₂CH₃ and C₁₆CO₂CH₃), 2.62 (s, 3H, N₆-CH₃), 2.09 (s, 3H, OCOCH₃), 1.06 and 0.64 (2t, $J_{18,19} = 7$, 6H, C₁₈-H₃ and C₁₉-H₃).

Preparation of 5'-nor-anhydrovinblastine 10a. Compound 4 (345 mg, 0.42 mmol) in soln in dry CH₂Cl₂ (2 ml) was treated at 0° under argon for 2 hr with trifluoroacetic anhydride (0.267 ml, 1.8 mmol). The medium was evaporated *in vacuo* without heating and dissolved in dry THF (8 ml). Water (0.03 ml) was introduced and the resulting soln was stirred for an additional 1 hr under argon at room temp, poured into brine, extracted with CHCl₃, washed with brine until neutral. The resulting organic layer was dried over Na₂SO₄, filtered and evaporated *in vacuo*. The crude residue was purified by preparative tlc (eluent: CHCl₃-MeOH, 90-10) and 10a (90 mg, 27%) and polar products (257 mg) were isolated.

The polar products were dissolved in a tetrahydrofuran-trifluoroacetic acid mixture (10 ml/0.08 ml) and reduced with an excess of sodium cyanoborohydride. After neutralisation with NaHCO₃ and extraction with CHCl₃, the crude mixture was purified by preparative tlc (CHCl₃-MeOH, 90-10) and 2^{1b} (120 mg, 35%) and 3^{1a} (60 mg, 18%) were isolated.

5'-Nor-anhydrovinblastine 10a. $[\alpha]_D^{25} + 52.4^\circ$ (CHCl₃, $C = 0.3$); IR: 3460, 2965, 1745, 1620; UV: 215(3700), 268(11000), 282(9500), 293(7600), 310(4400); DC: 312 (+3.4), 305 (+3.2), 258 (+13.3), 230 (+31.6), 260 (-44.3); MS: 808.4420 (calc.: 808.4411, C₄₇H₄₆N₆O₈, 2%), 794.4235 (794.4254, C₄₆H₄₄N₆O₈, 4%), 792.4897 (792.4098, C₄₆H₄₄N₆O₈, 11.5%), 748.4155 (748.4200, C₄₅H₄₂N₆O₈, 100%), 734.4001 (734.4043, C₄₄H₄₀N₆O₈, 23%), 598.2909 (598.2917, C₄₃H₃₈N₆O₈, 70%), 282.1335 (282.1341, C₁₄H₁₂NO₅, 8.2%), 135.1048 (135.1048, C₆H₁₁N, 8%), 122.0976 (122.0970, C₆H₁₁N, 3%). NMR ¹H (250 MHz): (s, 1H, N₆-H), 7.17 (m, 2H, H aromatic indolic), 6.34 and 6.08 (2s, 2H, C₉-H and C₁₇-H), 5.83 (dd, 1H, $J_{14,15} = 9.5$ and $J_{3,11} = 3.5$, 1H, C₁₅-H), 5.76 (m, 1H, C₁₅-H), 5.37 (s, 1H, C₁₇-H), 5.25 (d, $J_{14,15} = 9.5$, 1H, C₁₅-H), 4.58 and 4.41 (d not well resolved and d, $J_{4,5,6} = 14$, 2H, C₄-H₂), 3.82, 3.13 and 3.68 (3s, 9H, C₁₁-OCH₃, C₁₆-CO₂CH₃ and C₁₈-CO₂CH₃), 2.69 (s, 3H, N₆-CH₃), 2.57 (s, 1H, C₂₁-H), 2.06 (s, 3H, OCOCH₃), 1.07 and 0.20 (2t, $J_{18,19} = 7$, 6H, C₁₈-H₃ and C₁₉-H₃). NMR ¹³C: 173.9 (C₁₆-COO), 171.3 and 170.6 (-OCOCH₃ and C₁₆COO), 157.9 (C₁₁), 152.7 (C₁₃), 134.2 (C₁₇ and C₂₀), 132.5 (C₇), 129.6 (C₁₄), 128.3 (C₈), 124.8 (C₁₄), 122.7 and 120.6 (C₁₀ and C₁₁), 118.7 (C₇), 118.4 (C₉), 110.4 (C₁₂), 93.8 (C₁₂), 83.1 (C₂), 79.8 (C₁₄), 64.7 (C₂₁), 55.8 (C₁₁-OCH₃), 54.7 (C₁₆), 53.6 and 53.4 (C₆ and C₇), 52.9 and 52.1 (C₁₆-CO₂CH₃ and C₁₈-CO₂CH₃), 50.3 and 49.8 (C₃ and C₄), 47.0 and 44.1 (C₂₁ and C₂), 44.6 (C₄) (attributed with respect to the resolution for this signal), 42.8 (C₂₀), 38.3 (N₆-CH₃), 35.2 and 27.7 (C₁₇ and C₁₉), 30.8 (C₁₉), 28.2 (C₁₄), 21.3 (COCH₃), 12.2 (C₁₈), 8.3 (C₁₈).

Preparation of 5'-nor-anhydrovinblastine 10a from compound 7b. Compound 7b (80 mg, 0.095 mmol) in dry THF (3 ml) was treated in the dark under argon at room temp. by silver tetrafluoroborate (38 mg, 0.2 mmol). The medium was stirred for 3 hr and then diluted with trifluoroacetic acid (0.12 ml) and water (0.12 ml). The resulting soln was stirred for an additional 1 hr, poured into brine and extracted with CHCl₃. The organic layers were washed with water until neutral. The crude mixture obtained after usual treatment was purified by preparative tlc (CHCl₃-MeOH: 92-8), and a product identical ($[\alpha]_D^{25}$, IR, UV, SM, ¹H NMR and HPLC with 10a) was obtained (50 mg, 68%).

Preparation of compound 13 from 15'-nor-anhydrovinblastine 10a. A soln of 10a (5 mg, 0.006 mmol) in dry MeOH (0.5 ml) was treated, at room temp. under argon with Ac₂O (0.02 ml) in the presence of KOAc (5 mg). The medium was stirred for 30 min, poured into brine and extracted with CHCl₃. After the usual treatment, the crude product was purified by preparative tlc (AcOEt-MeOH: 90-10) and compound 13 was isolated (3 mg, 55%).

Compound 13. IR: 3420, 2920, 1740, 1640; UV: 220, 226, 284, 292, 314; SM: 852(M⁺), 821, 692, 661(100%), 553, 497, 496, 380, 282, 273, 223, 135, 122, 121; ¹H NMR (240 MHz): 6.0 and 5.5 (2s, 2H, C₉-H and C₁₇-H), 3.80 (s), 3.72 and 3.70 (2s), 3.44 and 3.40 (2s) (9H, C₁₁CO₂CH₃, C₁₆-CO₂CH₃ and C₁₁-OCH₃), 3.14 and 3.06 (2s) (3H, C₂-OCH₃), 2.70 (s, 3H, N₆-CH₃), 2.10 (s, 3H, OCOCH₃), 1.90 (s, 3H, N₆-COCH₃).

Preparation of 5'-nor-anhydrovinblastine N₆-oxide. Compound 10a (10 mg, 0.013 mmol) in soln in CH₂Cl₂ (5 ml) was treated at 0° for 10 min with 3-chloroperbenzoic acid (3 mg, 0.017 mmol). The medium was poured into CHCl₃, washed with Na₂CO₃ aq (40 g/l) and 5'-nor anhydrovinblastine N₆-oxide was obtained (quantitative) after the usual treatment.

SM: 808, 794, 792, 776, 748, 656, 589(100%), 538, 282, 152, 135, 122, 121; ¹H RMN (400 MHz): 8.56 (s, 1H, N₆-H), 7.91 (1H, C₉-H), 6.65 and 6.1 (2s, 2H, C₉-H and C₁₇-H), 5.93 (dd, 1H, C₁₅-H), 5.67 (m, 1H, C₁₅-H), 5.36 (s, 1H, C₁₇-H), 5.27 (d, 1H, C₁₅-H), 4.91 and 4.68 (2d, $J_{4,5,6} = 13$, 2H, C₄-H₂), 3.85, 3.78 and 3.71 (3s, 9H, OCH₃, C₁₆-CO₂CH₃ and C₁₈-CO₂CH₃), 2.73 (s, 3H, N₆-CH₃), 2.1 (s, 3H, OCOCH₃), 1.08 and 0.70 (t, 3H, $J_{18,19} = 7$, C₁₈-H₃ and C₁₉-H₃).

Acknowledgements—The authors are grateful to Dr. P. Bladon (University of Strathclyde, Glasgow, Scotland) for high resolution mass measurements of 5'-nor anhydrovinblastine 10a, they are also indebted to the Ligue Nationale Française contre le Cancer for a grant to R. Z. Andriamialison and to Dr. S. K. Kan and his colleagues for 240 and 400 MHz ¹H NMR spectroscopy.

This work is dedicated to the memory of the late Prof. J. Le Men deceased 4 October 1978.

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